

Defect-Driven Magnetism in Luminescent *n/p*-Type Pristine and Gd-Substituted SnO₂ Nanocrystalline Thin Films

S. Ghosh,* Gobinda Gopal Khan, and K. Mandal

Department of Condensed Matter Physics and Material Sciences, S. N. Bose National Centre for Basic Sciences, Block JD, Sector III, Salt Lake City, Kolkata 700 098, India

ABSTRACT: The effects of rare-earth-element Gd doping on the intrinsic magnetic ordering, photoluminescence, and electrical-conducting properties of the pristine SnO₂ nanocrystalline thin films fabricated by radio-frequency (RF) sputtering are investigated. The pristine SnO₂ thin film exhibits significant ferromagnetism while Gd doping results in an absence of intrinsic ferromagnetism. The presence of large amounts of singly ionized oxygen vacancies (V_{O}^+) is traced by photoluminescence spectroscopic analysis and they are found to be responsible for the observed ferromagnetism in pristine SnO₂ thin films. A significant reduction of oxygen vacancies is observed after Gd doping, and that might be insufficient to mediate long-range ferromagnetic ordering between V_{O}^+ defects in a Gd-doped SnO₂ system. Although the associated magnetic moment is increased by 1 order of magnitude, because of the insertion of Gd³⁺ ions, which have localized *f*-shell paramagnetic moment, there is no intrinsic FM ordering. Hall measurement reveals that the pure SnO₂ exhibits *n*-type behavior whereas Gd-doped SnO₂ films show the *p*-type conductivity with higher resistivity. The studies demonstrate that only structural defects such as V_{O}^+ defects, not magnetic ions such as Gd³⁺, are responsible for inducing ferromagnetism in SnO₂ thin films.

KEYWORDS: tin dioxide, ferromagnetism, structural defects, photoluminescence, conductivity

1. INTRODUCTION

The issue of magnetism in oxide semiconductor thin films has attracted immense attraction for current research, because of their considerable practical application in spin- and charge-controlled devices, which are known as spintronics.^{1,2} In 2000, Dietl et al.³ demonstrated that ferromagnetic mediation between Mn ions in *p*-type Mn-doped GaAs is possible and they have predicted the possibility of high Curie temperature (T_c) ferromagnetism (FM) in oxide and nitride semiconductors. Since then, large numbers of reports have appeared in the literature demanding room-temperature ferromagnetism (RTFM) in transition-metal (TM)- or rare-earth-element (RE)-doped wide-band-gap oxide semiconductors.^{4–7} But still, the origin of FM in such magnetic ion-doped oxides has been greatly challenged by some scientists and eventually leads to a controversial issue about the presence or absence of FM.^{8–12} Ogale et al.¹³ reported FM conditions in a Co-doped SnO₂ thin film that had a giant magnetic moment ($\sim 7.5 \pm 0.5 \mu_B$ per Co ion). Liu et al.¹⁴ have shown that FM in Co-doped SnO₂ nanocrystalline thin films is mediated by structural defects and magnetization is dependent on the thickness of films. In our earlier work, we found that Co-doped SnO₂ exhibits paramagnetism (PM) in bulk systems, whereas non-DMS-related FM is observed in Co-doped SnO₂ nanorods.¹⁵ Recently, research also has shown the absence of FM in TM-doped oxides, with their high structural perfection.^{16,17} Fernandes et al.¹⁸ also have observed the significant loss of the ferromagnetic signature in the CeO₂ films upon the low level doping of the Mn, Fe, Co, and Cu ions in the matrix. Therefore, considering the controversy related to the issue of FM in magnetic ion-doped oxides, the RTFM in pure and nonmagnetic-ion-doped oxides has been investigated.^{19–21} The reported RTFM in pure HfO₂ thin films by Venkatesan et al.¹⁹

has opened up a new phenomenon, called d^0 -ferromagnetism. RTFM has also been reported for pure oxides such as ZnO, SnO₂, and TiO₂.²⁰ The *ab initio* calculations performed by Pemmaraju et al.²² for HfO₂ and by Elfimov et al.²³ for CaO have indicated that the cationic vacancies can induce an almost localized magnetic moment on the oxygen atoms neighboring the vacancy site. In our earlier work, we also have found that the FM in ZnO nanowires can be tuned by controlling Zn vacancy (V_{Zn}) defects using nonmagnetic potassium (K) as a dopant.²¹ On the other hand, oxygen vacancies (V_{O}) have also been attributed to introducing RTFM in wide-band-gap oxide semiconductors such as ZnO, In₂O₃, and SnO₂.^{20,24,25} The controlling of the FM by tuning of the oxygen vacancies is reported in the Li–Ni-codoped ZnO,²⁶ ZnO,²⁷ and Cr-doped In₂O₃^{28,29} thin films and nanostructures.

SnO₂ is an important *n*-type semiconductor with remarkable optical properties, which are due to its wide band gap (~ 3.6 eV). During the preparation of SnO₂ thin films under variable conditions, different intrinsic structural defects such as Sn interstitial (Sn_i), oxygen vacancy (V_{O}), and Sn vacancy (V_{Sn}) can be generated. Among them, only the Sn vacancy (V_{Sn}) and singly ionized oxygen vacancy (V_{O}^+) are reported to show localized moments, whereas the other defects (such as Sn_i or neutral V_{O} or V_{O}^{2+}) are nonmagnetic.^{30–32} The density functional theory (DFT) study performed by Rahman et al.³¹ showed that V_{Sn} can generate a large magnetic moment around $4 \mu_B$ and they also have attributed the experimentally observed giant magnetic moment of Co-doped SnO₂ thin films¹³ to the V_{Sn} defects. Generally, it is very unlikely to have a considerable

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amount of V_{Sn} defects in pure SnO_2 thin films due to the high formation energy of V_{Sn} defects. On the other hand, oxygen vacancy (V_{O}) can easily grow during SnO_2 thin film deposition, because of its low formation energy.³³ However, there are limited numbers of reports about the defect magnetism in SnO_2 thin films, unlike ZnO. In fact, the effects of the rare-earth-element (RE)-ion doping on the magnetic properties of the SnO_2 thin films are still limited.³⁴ The doping of the RE element in the SnO_2 matrix can provide the exciting optical and magnetic properties of the thin films. Hence, RE-ion-doped SnO_2 could be interesting from the viewpoint of optical and spintronic applications. Moreover, trivalent Gd ions have larger localized magnetic moments, because of their $4f^7$ electron configuration, compared to the TM ions. Therefore, it is expected that the Gd-doped oxides could exhibit a large ferromagnetic moment and the substitution of trivalent Gd ions at the tetravalent Sn site can also introduce holes in the SnO_2 system, which can be favorable to sustaining FM coupling. Therefore, in this report, we have investigated the origin of FM in pristine and Gd-doped SnO_2 thin films coupled with their magnetic, photoluminescence, and electrical properties. We have found that singly ionized oxygen vacancies (V_{O}^+) are responsible for inducing FM in SnO_2 nanocrystalline thin films, whereas the Gd doping has little influence on FM in the SnO_2 thin films.

2. EXPERIMENTAL DETAILS

Pure and Gd-doped SnO_2 thin films have been deposited on a Si(100) substrate using a radio-frequency (RF) sputtering technique. For the preparation of the target, high-purity SnO_2 (99.99%) and Gd_2O_3 (99.99%), in the appropriate ratio, are mechanically milled using Fritsch Pulverisette-7 mill. The mixture is sintered at 800 °C for 12 h and a pallet 1 in. in diameter is prepared from it. The pallet is sintered again at high temperature (1000 °C) for 4 h. Afterward, it is used as target material for the deposition of SnO_2 and $\text{Sn}_{1-x}\text{Gd}_x\text{O}_2$ ($x = 0.02$ and 0.05) thin films. The RF sputtering is conducted using a power of 50 W dc, where the ratio of argon (Ar) and oxygen (O_2) is 10:1. The base pressure of the sputtering chamber is 2.7×10^{-2} mbar. The average deposition rate is ~ 4.5 nm/min, and the total deposition time is 45 min. Afterward, the as-deposited pure and Gd-doped SnO_2 films are annealed at 700 °C in an oxygen atmosphere for 3 h to improve the crystalline quality of the films. All the annealed SnO_2 thin films are characterized by XRD for phase identification and thicknesses of the films are measured using an ellipsometer. Scanning electron microscopy (SEM) is used to examine the surface morphology of the films. Magnetic measurements are performed by superconducting quantum interference devices (SQUIDs). Photoluminescence (PL) spectroscopic measurements are conducted to observe the defect-levels transitions between the optical band gap of SnO_2 thin films. Hall measurement is carried out to investigate the types (n -type/ p -type) of the semiconductors and to estimate the carrier concentrations. The resistivity of films is measured with standard four-probe method with two current and voltage leads.

3. RESULTS AND DISCUSSION

The crystallographic phase identification of the pure and Gd-doped SnO_2 thin films have been performed using XRD (patterns are shown in Figure 1). All the peaks with the mentioned (hkl) values correspond to the tetragonal rutile structure of the SnO_2 lattice. Note that all the films are single phase, without the presence of any impurity phase. The peak at $\sim 44.3^\circ$ originates from the Si substrate upon which the SnO_2 films are deposited. The variation of the lattice parameters and d -spacing (d_{110} and d_{101}) between parallel planes of SnO_2 unit cell, estimated using Bragg's diffraction law, are shown in

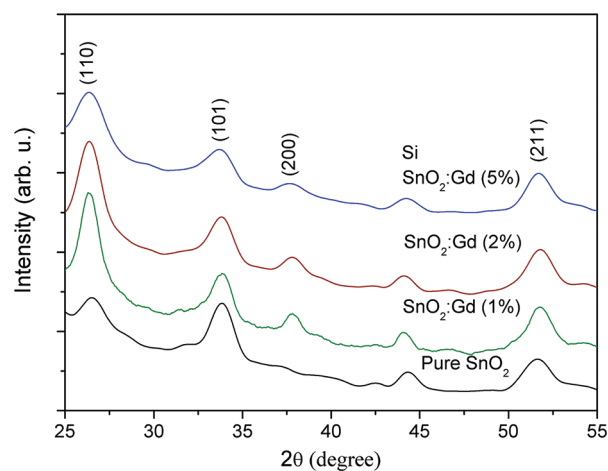


Figure 1. X-ray diffraction (XRD) patterns of pristine SnO_2 and Gd-doped SnO_2 thin films at room temperature (300 K).

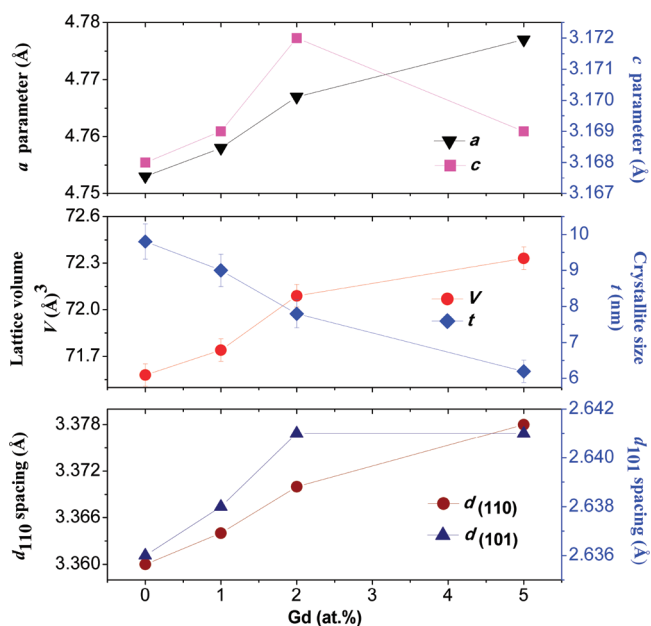


Figure 2. Variation of lattice parameters (a and c), lattice volume (V), crystallite size (t), and d -spacing of the (110) and (101) crystal planes of SnO_2 lattice, relative to Gd concentration (at.%).

Figure 2. It is important to notice that the lattice parameters (a ($a = b$) and c) as well as lattice volume (V) are found to increase with Gd doping. This expansion of the lattice volume ensures the substitution of larger Gd^{3+} ions (1.07 Å) replacing smaller Sn^{4+} ions (0.69 Å) in the SnO_2 matrix. This type of variation of the lattice parameters is also observed for other TM-doped SnO_2 thin films.^{10,11,14} The values of lattice parameters and d -spacing for different crystallographic planes of the different films are also summarized in Table 1. The full width at half-maximum (fwhm) of the (110) peak is observed to increase with increasing Gd doping, which signifies a decrease in the average crystallite sizes (t) of the films. The average crystallite sizes of the thin films are estimated using the Debye–Scherrer formula,

$$t = \frac{0.9\lambda}{\beta \cos \theta}$$

Table 1. Typical Estimated Values of Lattice Parameters (a and c), Lattice Volume (V), d -Spacing of the (110) and (101) Planes, Full Width at Half Maximum (fwhm) of the (110) Plane, and Crystallite Size (t) of As-Deposited Pristine and Gd-Doped SnO₂ Thin Films

thin film	Lattice Parameters			d -Spacing (Å)		fwhm at (110) (°)	crystallite size, t (nm)
	a^a (Å)	c (Å)	volume, V (Å ³)	d_{110}	d_{101}		
SnO ₂	4.753	3.168	71.56 ± 0.07	3.360	2.636	1.85 ± 0.02	9.8 ± 0.3
SnO ₂ :Gd (1%)	4.758	3.169	71.74 ± 0.07	3.364	2.638	2.02 ± 0.02	9.0 ± 0.3
SnO ₂ :Gd (2%)	4.767	3.172	72.08 ± 0.07	3.370	2.640	2.34 ± 0.02	7.8 ± 0.3
SnO ₂ :Gd (5%)	4.777	3.169	72.31 ± 0.07	3.378	2.641	2.91 ± 0.02	6.3 ± 0.3

^aNote that $a = b$.

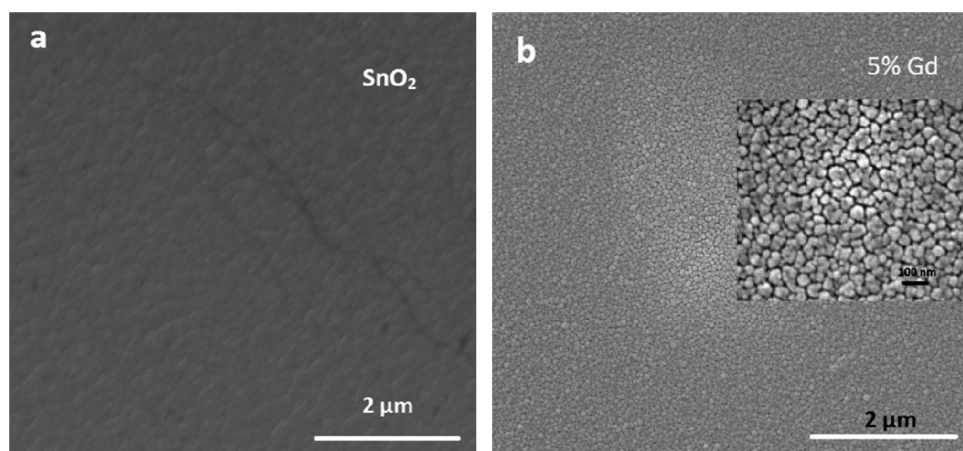


Figure 3. SEM images of (a) a pristine SnO₂ thin film and (b) a 5% Gd-doped SnO₂ thin film. Inset in panel b shows a high-magnification image of the corresponding thin film, revealing the nanocrystalline nature of the film.

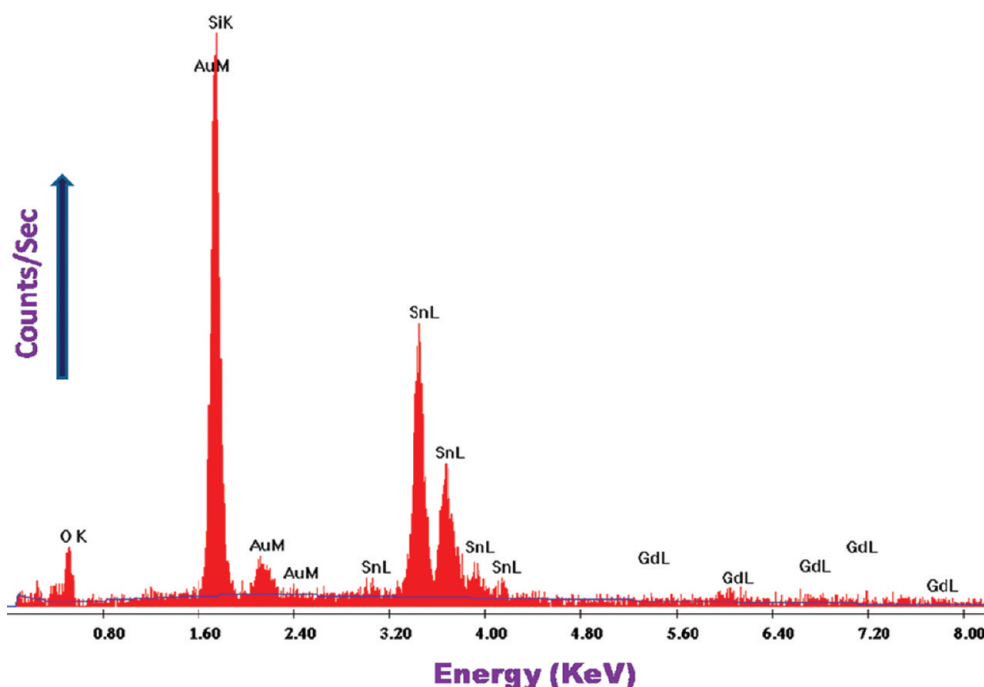


Figure 4. Energy-dispersive X-ray (EDAX) spectrum of the 5% Gd-doped SnO₂ thin film.

where λ is the wavelength of X-ray beam used (1.54 Å for Cu $K\alpha$ radiation), θ the peak position of the given (hkl) peak, and β the corresponding fwhm in radian. The average t values of all thin films are found to be <10 nm, which demonstrates the nanocrystalline nature of the thin films (see Table 1). The thicknesses of the thin films measured by ellipsometer are

found to be 228, 217, 210, and 205 nm for pure, 1%, 2%, and 5% Gd-doped SnO₂, respectively.

Figure 3 shows the low-magnification SEM images of the pure SnO₂ thin film (Figure 3a) and the 5% Gd-doped SnO₂ thin film (Figure 3b); the inset of Figure 3b shows the high-magnification image of the corresponding thin film. The SEM

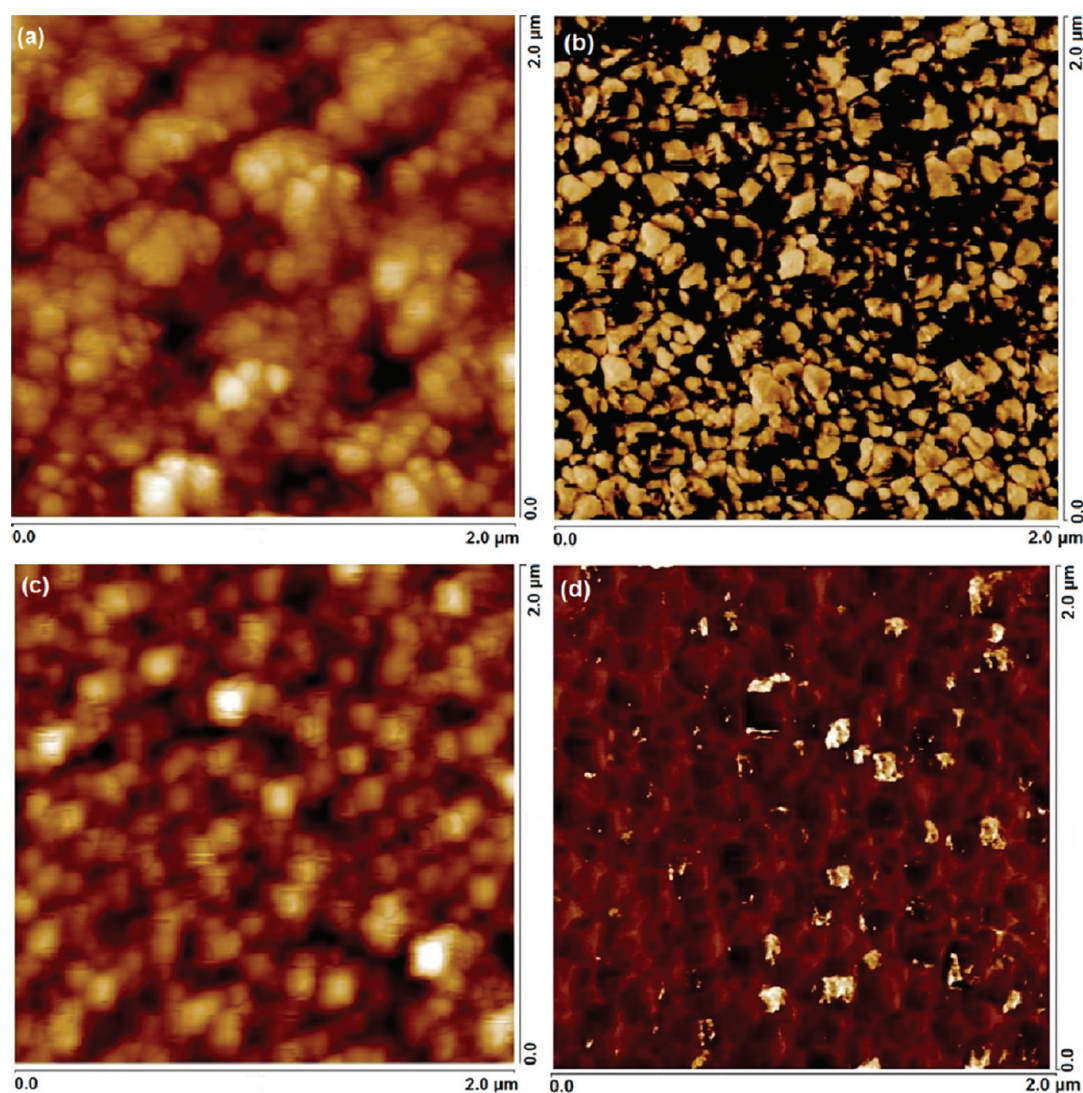


Figure 5. (a, c) Topographic atomic force microscopy (AFM) images and (b, d) the corresponding grainlike magnetic domain structures revealed by the magnetic force microscopy (MFM) imaging, for the pure film (panels a and b) and the 5% Gd-doped SnO₂ film (panels c and d).

images show that both the pure and Gd-doped SnO₂ films are homogeneous throughout, demanding a uniform Gd doping with no accumulation of Gd for the Gd-doped film. From the SEM image of the 5% Gd-doped SnO₂ thin film (inset of Figure 3b), it is also clear that the thin film is nanocrystalline, with prominent grains having an average grain size of ~50 nm. The grains are also very clearly discernable for the Gd-doped SnO₂ thin films, compared to that of pure SnO₂ thin films, which indicates a better crystallinity of the 5% Gd-doped SnO₂ film. This is probably because the doping of trivalent Gd ions, which favors the achievement of better crystallinity in SnO₂ films. The energy-dispersive X-ray (EDAX) spectrum of the 5% Gd-doped SnO₂ thin films is shown in Figure 4. The Gd concentrations in the thin films quantified by EDAX spectroscopy are found to be 4.87, 1.91, and 0.82 at.% for nominal Gd concentrations of 5, 2, and 1 at.%. The Si K peak originates from the Si substrate on which SnO₂ films are deposited, whereas the Au K signal appears from the gold layer coated on the SnO₂ films (which was required for the SEM investigation).

Figure 5 shows the atomic force microscopy (AFM) and the corresponding magnetic force microscopy (MFM) images of the pure and 5% Gd-doped SnO₂ films at a lift height of

~300 nm over a scan size of 2 μm × 2 μm. It is evident from Figure 5 that the MFM features of the films are different from the topography of the films. It is found that the MFM amplitude image shows some coarse grainlike domain structures, where the bright and dark contrast of domains correspond to high concentrations of positive and negative poles, respectively. The domain configurations are also different for the pure and 5% Gd-doped SnO₂ films. The grainlike domain structures are more prominent and vivid in the case of the pure SnO₂ nanocrystalline thin films (see Figure 5b), compared to the 5% Gd-doped SnO₂ film (see Figure 5d). In the case of 5% Gd-doped SnO₂ films, the bright grainlike domains are less in number, and they are random and far apart from each other. Therefore, the observed regular distribution of the bright and dark grainlike magnetic domains in the pure SnO₂ thin film is an indication of long-range magnetic ordering in it over the Gd-doped SnO₂ films.^{35–37}

Figure 6 shows the field dependent magnetization M (H) measurements of pure and Gd-doped SnO₂ thin films. Interestingly, the pure SnO₂ thin film exhibits a clear hysteresis loop, which indicates that the pristine SnO₂ thin films are ferromagnetic at room temperature (Figure 6a) as well as at

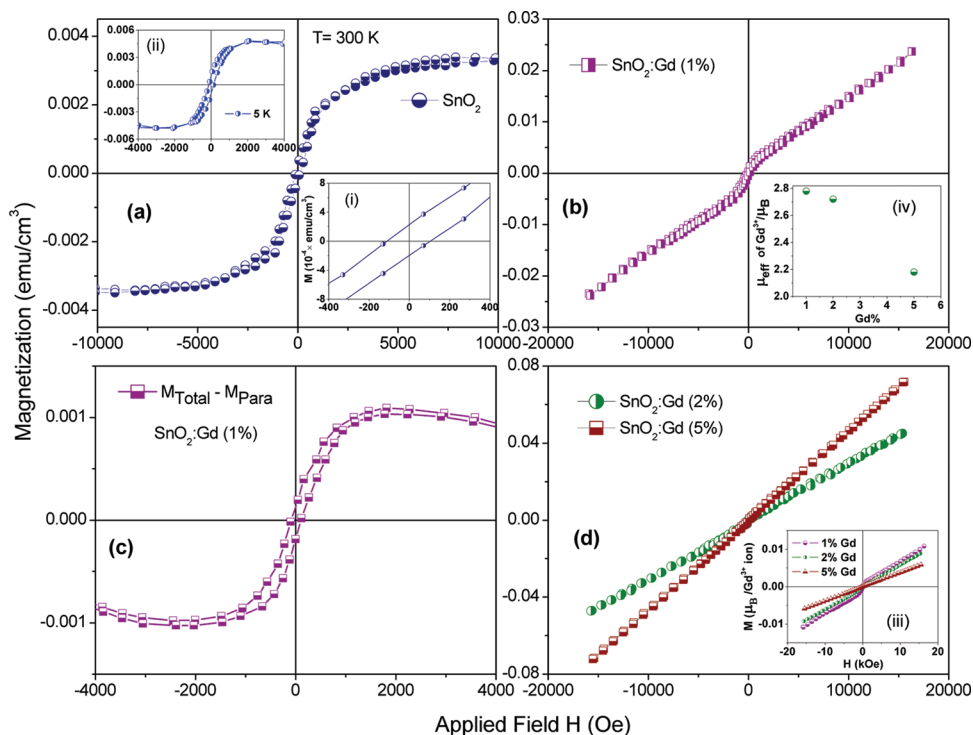


Figure 6. Field dependent magnetization ($M(H)$) loops of (a) the pristine SnO_2 thin film at 300 K, (b) 1% Gd-doped SnO_2 , (c) the ferromagnetic component ($M_{\text{Total}} - M_{\text{Para}}$) of the 1% Gd-doped film after subtracting the paramagnetic component and (d) the 2% and 5% Gd-doped SnO_2 thin films. Insets show (i) the magnetization of pure SnO_2 in low field area at 300 K, (ii) the $M-H$ loop of pure SnO_2 at 5 K, (iii) the magnetization of Gd-doped films (in units of μ_{B} per Gd^{3+} ion), and (iv) the effective magnetic moment (μ_{eff}) of the individual Gd ion in units of Bohr magneton (μ_{B}) versus the Gd concentration in the film. (All data presented after correcting the diamagnetic substrate contribution.)

low temperature (inset (ii) of Figure 6a). A significant amount of coercivity ($H_{\text{c}} \approx 110$ and 125 Oe at 300 and 5 K, respectively) is observed in the case of pure SnO_2 film, which ensures the FM nature. The saturation magnetization (M_{s}) and retentivity (M_{r}) are found to be 3.4×10^{-3} and 2.2×10^{-4} emu/cm^3 , respectively, at 300 K and 4.6×10^{-3} and 9.8×10^{-4} emu/cm^3 , respectively, at 5 K. This type of ferromagnetic response in other pristine oxides has also been reported earlier.^{19–21} On the other hand, it is quite interesting to observe that Gd-doped SnO_2 films show a linear $M-H$ behavior (see Figures 6b and d), which indicates that a dominant paramagnetic ordering is present in the Gd-doped films. In the case of the 1% Gd-doped film, a curvature is observed near the origin (Figure 6b), which indicates that a ferromagnetic component is superimposed over a strong paramagnetic background. Therefore, we have subtracted the strong paramagnetic component (M_{Para}) from the total moment (M_{Total}) and plotted the residual component ($M_{\text{Total}} - M_{\text{Para}}$) in Figure 6c, which shows the clear hysteresis-type behavior with a coercivity of $H_{\text{C}} \approx 95$ Oe. The saturation magnetization (M_{s}) is determined to be 1.05×10^{-3} emu/cm^3 , which is much smaller, compared to that of the pure SnO_2 film. Therefore, with 1 at.% Gd doping in the SnO_2 film, the FM is found to be suppressed subsequently and a large paramagnetic moment is established within the system. For the 2% Gd-doped film, the ferromagnetic component ($M_{\text{Total}} - M_{\text{Para}}$) is observed to have completely vanished, while the paramagnetic component has been enhanced significantly. Similarly, no FM is found for 5% Gd-doped films either. Therefore, the FM in pure SnO_2 film tends to diminish gradually with the increase in Gd doping, and then, over a certain Gd concentration (lying somewhere between 1 and 2 at.%), it completely vanishes. However, the earlier MF

studies also showed an indication of long-range ferromagnetic ordering in the pure SnO_2 films, whereas no such observation was found for the 5% Gd-doped film. Note that, although the associated magnetic moment is increased by 1 order of magnitude, because of the insertion of Gd^{3+} ions, which have a localized f -shell paramagnetic moment, there is no intrinsic ferromagnetic ordering in the Gd-doped SnO_2 films. However, similar results were also obtained by Ney et al.,¹⁷ Xu et al.,³⁸ and Barla et al.³⁹ in the case Co-doped ZnO thin films and also in our previous work on the Co-doped SnO_2 .¹⁵ Recently, a loss of magnetization is also observed in CeO_2 thin films¹⁸ with the doping of TM ions. Herein, we estimated the magnetization per Gd^{3+} ion and the effective magnetic moment (μ_{eff}) of the individual Gd ion in units of Bohr magneton (μ_{B}), shown in insets (iii) and (iv) of Figure 6, respectively. Note that the magnetization per Gd ion, as well as the μ_{eff} value, decreases as the Gd concentration increases, which may be due to enhancement of the antiferromagnetic superexchange interaction between Gd^{3+} ions via oxygen ligands. With the increase of Gd concentration in the SnO_2 matrix, the average distance between the Gd^{3+} ions should decrease, leading to the enhancement of this type of antiferromagnetic superexchange interaction. The μ_{eff} value of the Gd^{3+} ion is found to be maximum ($2.78 \mu_{\text{B}}$) in 1% Gd-doped SnO_2 film, which is much less than the full moment of the Gd^{3+} ion ($7.94 \mu_{\text{B}}$). Similar results are also observed in the case of Co: ZnO ,¹⁰ Co: SnO_2 ,¹⁵ and Gd: ZnO ,⁴⁰ where the effective moment of the respective TM or RE ion is observed to decrease significantly. This can happen because the moment of the magnetic ions is dependent on the ionic environment^{41,42} and ion symmetry within the host matrix. The increase of Gd concentration results in a lattice distortion, which can change the Gd^{3+} site

symmetry as well as the effective magnetic moment of the Gd^{3+} ion.

To understand the magnetic properties in further detail, zero-field-cooled (ZFC) and field-cooled (FC) measurements are also performed for Gd-doped SnO_2 films, shown in Figure 7.

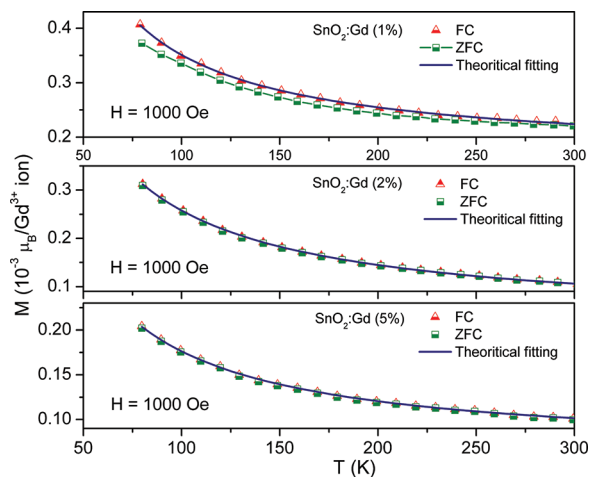


Figure 7. Field-cooled/zero-field-cooled (FC–ZFC) plots for all the Gd-doped SnO_2 thin films in a constant field of 1000 Oe, along with the theoretical fitting of FC curves according to the modified Curie–Weiss law.

In the case of the 2% and 5% Gd-doped SnO_2 films, both ZFC and FC curves (Figures 7b and 7c) trace the same path over the entire range of temperature, which demonstrates the presence of noninteracting paramagnetic moments in the system. However, in the case of the 1% Gd-doped film, the FC curve showed a higher moment than ZFC, which indicates the presence of a ferromagnetic phase. FC curves are also fitted according to the modified Curie–Weiss law:

$$\chi = \frac{m(T)}{H} = \chi_0 + \frac{C}{T + \theta}$$

where χ_0 represents weak nonparamagnetic susceptibility; C is the Curie constant ($C = N\mu^2/(3k)$, where N is the number of magnetic ions per gram, μ the magnetic moment of the ion, and k the Boltzmann constant), which represents the paramagnetic ion concentration; T is the temperature (in Kelvin); and θ is the Curie–Weiss temperature, which represents the magnetic exchange interactions between the localized spins. These fits yields θ values of 4.25, 3.8, and -9.67 K for 5%, 2%, and 1% Gd-doped SnO_2 films, respectively. Note that, unlike other films, the 1% Gd-doped film showed negative Curie–Weiss temperature (θ), which demonstrates the existence of a ferromagnetic exchange interaction in this film. The positive θ value for the 2% and 5% Gd-doped films signifies the existence of the antiferromagnetic exchange interaction between the Gd^{3+} ions. Gd^{3+} – O^{2-} – Gd^{3+} superexchange interaction might be the possible origin of the antiferromagnetism, whereas the Gd^{3+} – Gd^{3+} direct interaction is paramagnetic in nature.

Therefore, the study of the magnetic properties of the pure and Gd-doped SnO_2 thin films generates two questions:

- Do magnetic ions (such as Gd^{3+}) play any role in inducing FM in such oxides? and
- What makes pure SnO_2 thin film ferromagnetic?

To address those questions, it is essential to study the intrinsic defects that might be present in SnO_2 thin films. Hence, we have performed room-temperature (RT) photoluminescence (PL) measurements, as shown in Figure 8, to investigate the

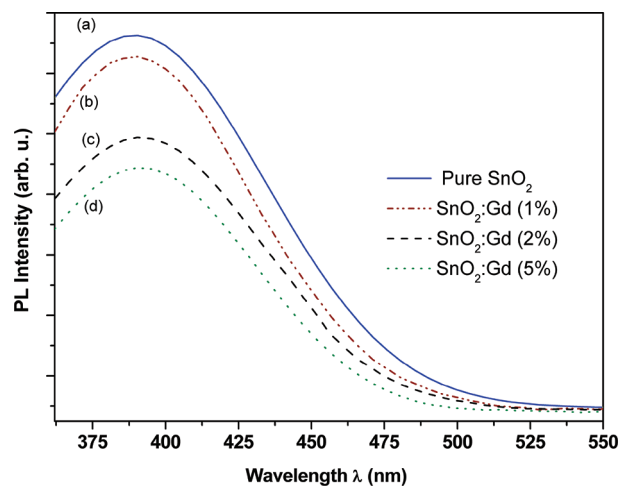


Figure 8. Photoluminescence (PL) spectrum of (a) pristine SnO_2 thin film and (b) 1%, (c) 2%, and (d) 5% Gd-doped SnO_2 thin films.

types of structural defects present in the pure and Gd-doped SnO_2 thin films. It is found that, under an excitation of 330-nm light from the Xe source, all the nanocrystalline thin films exhibit an intense broad PL emission band centered at 390 nm (3.17 eV). Since this emission energy (3.17 eV) is much lower than the energy band gap of SnO_2 (3.6 eV), the visible emission cannot be assigned to the direct recombination of conduction electron to the holes in valence band; instead, they are assigned to the defect-related emission. The strong blue emission from the thin films might be related to the inherent crystalline defects induced during their growth.^{43–45} It is important to note that the broad nature of the PL emission band clearly indicates that the luminescence should have originated from the multiple sources rather than a single source, because the PL band can be well-fitted with the multiple PL peaks. It is reported that the SnO_2 nanostructures can exhibit emission bands between 400 and 600 nm, because of the presence of different crystalline defects.^{46–48} The Sn vacancies (V_{Sn}) and oxygen vacancies (V_{O}) are the main structural defects present in the SnO_2 nanostructures responsible for introducing the defect states in the mid band gap, leading to luminescence in the visible region.⁴⁹ Although the formation energy of V_{Sn} is higher than that of V_{O} , it is expected that the concentration of V_{Sn} defects in the SnO_2 should be much lower than that of the oxygen vacancies.^{21,33} It has been reported that the nanocrystalline SnO_2 thin films exhibit intense photoemission at ~ 3.13 eV at different temperatures.⁵⁰ The blue luminescence from the SnO_2 thin films in the wavelength range of 400–450 nm has also been reported by other researchers,^{43,51} yet the origin of this PL emission band still has not been understood properly. It is assumed that the PL peaks might be attributed to the defects in the nanocrystal grains,^{43,50} or the oxygen vacancy defects,^{43,50,51} or even the Sn interstitial (Sn_i) defects.³⁹ However, the broad photoemission peak centered at ~ 400 nm is believed to originate because of the electron transition mediated by the V_{O} -related defects in the band gap.^{43,50,51} With the energy level of the V_{O} defects being highly controversial, in this experiment,

the PL emission peak centered at ~ 390 nm is attributed to the V_{O} -related defects, mainly singly ionized oxygen vacancy (F^+ center or V_{O}^+) defects, which are believed to be the recombination center for the luminescence.^{20,29} In our experiment, it is quite expected that the films should contain the V_{O}^+ defects, because the films are deposited in an oxygen-deficient atmosphere. However, here, we cannot also discard the presence of the V_{Sn} defects in the pure and Gd-doped SnO_2 films: the cation vacancy defects are also quite obvious in the lattice of the oxide semiconductors. Recently, the cation-vacancy-related defects are attributed to the visible emission as well as to the RTFM in the doped ZnO thin films and nanostructures.^{21,52–54} It has been found that the doping of the suitable element can also help in the formation of the cation vacancies in the lattice.²¹ It is evident from Figure 8 that the intensity of the PL emission centered at 390 nm is maximum for the pure SnO_2 thin film. Therefore, it is expected that the singly ionized oxygen vacancy (V_{O}^+) defect concentration is maximum in the pristine SnO_2 nanocrystalline film, compared to the Gd-doped films. However, the decrease of the intensity of the PL emission band with the increase of the Gd concentration in the Gd-doped SnO_2 films may be attributed to the considerable attenuation of the oxygen vacancies in the lattice, where the V_{Sn} defects could predominate and the PL emission bands becomes broader in nature with the increase of the Gd concentration in the matrix.

Here, it is important to mention that enough precaution is taken during the preparation of the films to avoid any type of contamination. Nonmagnetic tweezers that have plastic tips are used to transfer the films during different measurements. In fact, no impurities have been detected from XRD and EDAX analysis considering their detection limit. Moreover, we also have tried to exclude any type of instrumental error by conducting the magnetic measurements repeatedly. Therefore, to discuss the origin of FM in pristine SnO_2 thin film, it is quite clear from the PL studies that intrinsic defects are playing the key role to induce FM in SnO_2 . The present PL analysis has shown that structural defects, such as V_{O}^+ defects, are maximum in the pure SnO_2 thin film. Here, it is worth mentioning that a single V_{O}^+ defect can generate a localized magnetic moment of $\sim 1.0 \mu_{\text{B}}$ and when this type of defect (V_{O}^+) concentration goes above the percolation limit, a ferromagnetic exchange interaction can be stabilized,^{20,32,45,55} as observed in the pure SnO_2 thin film. On the other hand, the V_{O}^+ defect concentration was observed to decrease with increasing Gd doping in SnO_2 films. Although the V_{O}^+ defect concentration decreased in the 1% Gd-doped film, still the amount might be sufficient to mediate ferromagnetic interaction in this film. Therefore, the 1% Gd-doped film exhibits a weak FM signature, compared to that of the pure SnO_2 film. However, when the Gd concentration exceeds 2 at.%, the V_{O}^+ defect concentration might have fallen below the required percolation limit to sustain long-range ferromagnetic ordering between V_{O}^+ defects and, hence, the destruction of FM. However, the paramagnetic moments associated with the localized Gd^{3+} ions still exist and that is contributing to the total magnetic moments. Previously, we have seen, from the Curie–Weiss fittings, that the exchange interaction exists between Gd^{3+} ions are only the antiferromagnetic type at very low temperatures but not ferromagnetic one. Therefore, we demonstrate that the FM arises only because of the intrinsic structural defects, such as V_{O}^+ defects of SnO_2 lattice and, here, the magnetic ions (such as Gd^{3+}) have no influence in inducing FM in SnO_2 thin films. However, it is

important to mention that the Sn vacancy (V_{Sn}) defects are also reported to generate localized magnetic moments.^{30,31} The previous PL measurements indicate that there might have been some V_{Sn} defects present in the thin films, and they may be low in defect concentration because of their high formation energy.³³ Hence, it is expected that the concentration of the V_{Sn} defects present in the films should be far below the required percolation threshold to mediate long-range ferromagnetic ordering in both SnO_2 thin films, because it is not probable to have a considerable amount of V_{Sn} defects in SnO_2 without proper defect-engineering. The chance of the formation of V_{O}^+ defects is much easier, because of its low formation energy;³³ in addition, in this work, the formation of V_{O}^+ defects in SnO_2 thin films is quite possible, because all the thin films are deposited in an oxygen-deficient atmosphere, where the oxygen:argon ratio is 1:10.

The study of the electrical conductivity and the type of the charge carriers present in the oxide semiconductor thin films can also provide information about the intrinsic nature of the films. In this regard, here, we have investigated the types of charge carriers and the carrier concentrations by Hall measurements and also the electrical conductivity of the pure and Gd-doped SnO_2 thin films. The variations of the resistivity (ρ), carrier concentration (n), Hall coefficient (R_{H}), and mobility (μ) of the SnO_2 thin films, relative to the Gd-doping concentration, are shown in Figure 9, and their values are

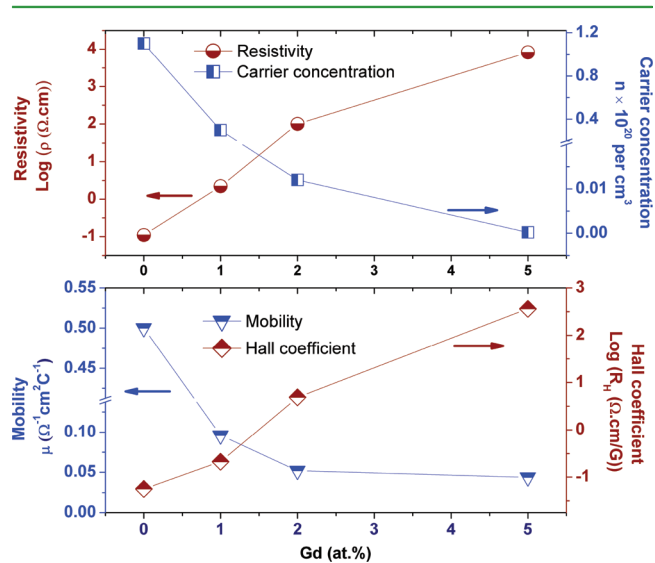


Figure 9. Variation of resistivity (ρ), carrier concentration (n), Hall coefficient (R_{H}), and mobility (μ) of SnO_2 thin films with Gd concentration.

summarized in Table 2. The negative value of R_{H} in the case of the pure SnO_2 and 1% Gd-doped thin film indicates n -type semiconducting nature, which is due to the presence of oxygen vacancies.^{33,56,57} On the other hand, the 2% and 5% Gd-doped SnO_2 films show a positive value of R_{H} , which indicates their p -type semiconducting behavior. This n -type to p -type transformation of the pure SnO_2 thin film with Gd doping can be understood because of the gradual substitution of Gd^{3+} ions at Sn^{4+} sites, which enhances hole concentration into the SnO_2 system. Figure 9 shows that, as the Gd-doping increases in the SnO_2 thin film, the electron concentration gradually decreases, which means an increase in hole concentration. The previous PL study also indicates that, with the increase of the Gd

Table 2. Typical Estimated Values of Resistivity (ρ), Hall Coefficient (R_H), Carrier Concentration (n), Mobility of Carriers (μ), and Types of Semiconductors for Pristine and Gd-Doped SnO₂ Thin Films

thin film	resistivity, ρ (Ω cm)	Hall coefficient, R_H (Ω cm/G)	carrier concentration, n (cm^{-3})	mobility, μ ($\Omega^{-1}\text{cm}^2\text{C}^{-1}$)	type of semiconductor
SnO ₂	1.1×10^{-1}	-5.6×10^{-2}	1.1×10^{20}	0.50	<i>n</i> -type
SnO ₂ :Gd (1%)	2.2	-2.1×10^{-1}	3.0×10^{19}	9.6×10^{-1}	<i>n</i> -type
SnO ₂ :Gd (2%)	10^2	+4.87	1.2×10^{18}	5.2×10^{-1}	<i>p</i> -type
SnO ₂ :Gd (5%)	8.3×10^3	$+3.6 \times 10^2$	1.7×10^{16}	4.4×10^{-1}	<i>p</i> -type

concentration in the SnO₂ film, the V_{Sn} defects could predominate as the concentration of V_{O} defects decreases significantly. The acceptor-type V_{Sn} defects may also contribute to the *p*-type nature of the doped SnO₂ thin films. The *n*-type conductivity of the pure SnO₂ thin film signifies that the film is enriched with the V_{O} defects, whereas the doping of the Gd³⁺ ion at the Sn⁴⁺ site can modify the multivalence of tin probably through the Sn_i population and, thus, the donor level inside the conduction band. As a result, the resistivity (ρ) of the pure SnO₂ thin film increases as the Gd-doping concentration increases. The 5% Gd-doped SnO₂ thin film becomes the most resistive one ($\rho = 8.3 \times 10^3 \Omega$ cm). The carrier concentration (n) is found to be on the order of $1.1 \times 10^{20} \text{ cm}^{-3}$ in the pure SnO₂ thin film, and it decreases monotonically as the Gd concentration increases. The mobility (μ) of the charge carriers also decreases with Gd doping and is quite expected, since the mobility of holes is less than that of electrons.

4. CONCLUDING REMARKS

In summary, we have successfully studied the role of structural defects in inducing ferromagnetism (FM) in pristine SnO₂ thin films and the effects of Gd doping on the magnetic, photoluminescence (PL), and electrical conduction of SnO₂ nanocrystalline thin films. Only the pristine SnO₂ thin film exhibits FM at both room temperature (RT) and low temperature; the FM signature in pure SnO₂ film is found to be suppressed initially with low Gd doping but is completely destroyed at higher dopant concentrations. The magnetic force microscopy (MFM) study indicates long-range magnetic ordering in pure SnO₂ films at RT. The PL spectra reveals the presence of large numbers of V_{O}^+ defects, which are found to be responsible for the observed FM in *n*-type pristine SnO₂ thin films. The studies indicate that the doping of Gd ions at Sn sites reduces the V_{O}^+ defect concentration significantly and makes the films highly resistive and *p*-type, whereas the pure SnO₂ film is highly conductive and *n*-type in nature. Although Gd ions introduce localized paramagnetic moments, still no long-range ferromagnetic ordering is observed in the Gd-doped SnO₂ thin films. The result signifies that magnetic ions such as Gd have little influence in inducing FM in SnO₂ thin films. Our work will certainly help to overcome the controversy about the origin of ferromagnetism in such magnetic ions-doped wide-band-gap oxide semiconductors.

AUTHOR INFORMATION

Corresponding Author

*Phone: +91 (033) 2335 5706-8. Fax: +91 (033) 2335 3477. E-mail: shyam@bose.res.in, sghoshphysics@gmail.com (S.G.).

Notes

The authors declare no competing financial interest.

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